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TRANSITIONAL METAL COMPLEXES OF NON STEROIDAL ANTI-INFLAMMATORY DRUGS

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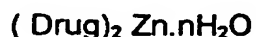
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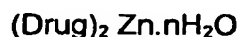
(57) Claim

1. A COX₂ selective pharmaceutical composition having potency ratio less than 1 comprising a complex of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -



wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent.

5. The use of an effective amount of a complex of Naproxen, its salt or adduct, or mixtures thereof, and zinc or a salt thereof, said complex represented by the formula -



wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent, in the preparation of a pharmaceutical composition for the treatment of an NSAID-indicated condition or symptom.

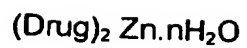
6. A method of treating an NSAID-indicated condition or symptom with reduced gastrointestinal side effects, said methods comprising:
administering an effective amount of a complex of Naproxen, its salt or adduct, or mixtures thereof, and zinc or a salt thereof, said complex

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represented by the formula -



wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent.

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ORIGINAL

COMPLETE SPECIFICATION

STANDARD PATENT

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Invention Title: "Transitional Metal Complexes of Non Steroidal Anti-Inflammatory
Drugs"

The following statement is a full description of this invention, including the best
method of performing it known to me:-

INTRODUCTION

The present invention relates to a novel pharmaceutical composition comprising Naproxen and a transition metal. The Novel drug metal complex is characterised in having increased efficacy and manifolds better tolerability than the parent drug and standard NSAIDS. More particularly the invention relates to pharmaceutical complex
5 of Naproxen and Zinc, obtained by reacting Naproxen and (or) one or more salts of Naproxen with Zinc in one or more salt form(s).

BACKGROUND OF THE INVENTION

Naproxen (US patents 3,904,682 & 4,009, 197) is (+) -2-(6-Methoxy-2-naphthyl) propionic acid [I.T. Harrison et al J. Med. Chem., 13 203 (1970)]. It has analgesic, anti-
10 inflammatory and antipyretic properties. It is an inhibitor of Cyclo-oxygenase [(Roszkowski et al Pharmacol. Exp. Ther 179, 114 (1971)), [Tomlinson et al Biochem Biophys Res Commun 46 552 (1972)]. Both Naproxen and Naproxen Sodium are used in musculoskeletal and joint disorders such as ankylosing spondylitis, osteoarthritis and rheumatoid arthritis, in mild to moderate pain such as dysmenorrhoea, migraine and
15 some musculoskeletal disorders, and in acute gout [R.N. Brogden et al Drugs, 18, 241-277 (1979)]. In some of the conditions the drug has to be administered for long periods of time. Among the other adverse-effects, Gastro-intestinal adverse effects are among the most frequently reported during short and long term treatment with Naproxen. Hence development of the pharmaceutical forms which reduce the adverse effects, in
20 particular the Gastro-intestinal adverse effects is most desirable.

There are several references of different types of approaches towards improvement of galenic dosage forms of Naproxen such as programmed release (US Patent Nos. 5480650, 4888178), Diffusion-osmotic controlled drug-release (US patent nos. 5543155, 4571333), Liquid-suspension controlled release (US Patent no. 5527545)
25 Systems for improved bio-availability or increased patient compliance e.g Naproxen micro capsules (W.O. Patent 9505166) and Cyclodextrin inclusion compounds (W.O

patent. 9504528) have also been reported. Soluble salts of Naproxen have been prepared with N-(2-hydroxy ethyl) pyrrolodone (U.S. patent no. 5206262).

Drug ligand-metal complexes of several therapeutic compounds have been reported. In several cases the ligand-metal complexes have better therapeutic and/or pharmaceutical properties than the parent ligand. The prior art regarding this aspect for NSAIDS is discussed below.

An inclusion complex of a NSAID or a pharmaceutically acceptable Salt thereof and a Cyclodextrin, and a physiologically acceptable alkali agent selected from the group consisting of alkali and alkaline earth metal carbonates, bicarbonates, phosphate and hydroxides, and water-soluble amines have been described in WO patent no. 9614839. The alkali agent has been shown to be capable of forming an alkaline diffusion layer around the composition in the gastro-intestinal tract.

Chen et al (J. Inorg. Biochem. 1992 Aug; 47 (2); 81-7) have described synthesis, characterisation, and anti-inflammatory activity of Naproxen complexes with rare earth (III). Various analyses have indicated presence of 1:3 (metal : ligand) stoichiometry and the carboxylate group of Naproxen has been suggested as a bridging ligand to co-ordinate to RE (III) ions. Two pharmacological models were chosen to examine the antiinflammatory activity of Nd(III) complex, which ascertained enhanced anti-inflammatory activity relative to the ligand.

It has been known now for 60 years that Zinc is essential for animal (including human) life. Approximately 300 enzymes require Zinc for their activities and it has a role in DNA Synthesis, Cell division and protein synthesis (Peasad A, Nutrition 1995, 11, 93-9). The inventors have looked into the possibility of making Zinc complexes of NSAIDs, particularly of Naproxen which have not been reported so far.

Navarro et al (Prostaglandins Leukot. Essent. Fatty Acids, 1994 Jun; 50 (6); 305-10) studied the effect of pre-treatment with Zinc acexamate on Gastrotoxic activities of dif-

ferent NSAIDs (diclofenac, indomethacin, Ketoprofen, naproxen and piroxicam). Zinc acexamate pre-treatment significantly decreased the overall Severity of lesions induced by NSAIDS. The experiments Corroborate the hypothesis that the preservation of the capability to synthesise endogenous prostaglandins is of critical importance in the maintenance of gastric mucosal integrity. The gastro-protective action observed with Zinc acexamate involves alternative mechanisms other than modification of PGE levels.

US Patent no. 5466824 describes a process for the preparation of a complex of indomethacin and a divalent metal comprising forming a solution by dissolving indomethacin and a salt of said divalent metal in a tertiary amide or cyclic tertiary amide, adding a C_{14} alkanol or C_{3-6} Ketone to the solution to precipitate the complex, and separating the precipitated complex from the solution. This art also provides a method for the treatment of inflammation and/or pain in a mammal requiring such treatment, comprising administering to said mammal an anti-inflammatory or analgesically effective amount of a complex of indomethacin and a divalent metal, the complex having the formula $(M_2)(Indomethacin)_n(S)_n$ wherein M is the divalent metal, S is a molecule of tertiary amide or a cyclic tertiary amide, and n is 2 or 3, or of a pharmaceutical composition comprising said complex together with a pharmaceutically acceptable carrier, diluent and/or excipient. This US patent further discloses a complex of indomethacin and a divalent metal, the complex having the formula $[M_2][indomethacin]_n[S]_n$, wherein M, S, and n are defined above, and a pharmaceutical composition comprising this complex together with a pharmaceutically acceptable carrier, diluent and/or excipient.

The zinc-aspirin complex has been reported by A.K. Singla et. al. (International Journal of Pharmaceutics, 108 (1994), 173-185). The complex was reported to have better efficacy and tolerability profile. Further Zinc-indomethacin complex has been reported (A.K. Singla et. al. International Journal of Pharmaceutics, 120 (1995) 145-155. The zinc-indomethacin complex was more potent than the parent drug and also had a better tolerability profile with respect to ulcerogenic properties.

U.S. Patent No. 5466824 describes a complex formation of indomethacin with a divalent metal and a tertiary amide or a cyclic tertiary amide while that to C_{14} alkanol or C_{3-6}

Ketone . Inclusion of alien molecules in a complex changes the total course of drug development process and substantial data in order to prove safety and efficacy of a new complex is essential. The process of preparation itself in U.S. Patent No. 5466824 is very cumbersome, time consuming, costly and the presence of new entities besides
5 the drug and the divalent metal may take considerable time and pre-clinical and clinical tests before the drug can be put to usefulness.

One of the inventors of the group of inventors of the present application has reported the formation of zinc-NSAID complexes. The zinc-indomethacin has been shown to be better than the parent molecule. Similarly, the aspirin-zinc complex has been shown to
10 be better than the parent molecule. However both these NSAIDs drugs have their own disadvantages in long term use. Indomethacin has low therapeutic index while aspirin has to be used in very high dosage to produce anaphylactic reactions. Although the zinc salts complexes are better, they may not be useful in therapeutics.

Besides indomethacin is not commonly used for therapy as an analgesic or antipyretic
15 because of high incidence and severity of side effects associated with long term administration. It is reserved for special use in acute gout and severe ankylosing spondylitis and osteoarthritis. This is evident from ED_{50} , LD_{50} and therapeutic index, values of NSAIDs as collected herein below.

S.No.	Drug	ED_{50} (mg/kg)	LD_{50} (mg/kg)	Therapeutic Index
20 1.	Nimesulide	1.25	324	260
2.	Naproxen	2.10	395	190
3.	Ibuprofen	13.5	923	68
4.	Diflumidone	38.0	750	20
5.	Flufenamic acid	14.7	249	17
25 6.	Phenylbutazone	20.5	406	14
7.	Acetylsalicylic acid	135.0	1520	14
8.	Indomethacin	2.95	210	7

Early approaches for improvement in efficacy and side effect profile were targetted with respect to drugs with low therapeutic index and poor safety margin.

NSAIDs can be ranked by concentration ($\mu\text{g/ml}$) of drug necessary to achieve 50% inhibition (IC_{50}) of COX_2 divided by IC_{50} for COX_1 , and the values below 1 indicate selectivity for COX_2 . The similarities in rates of g.i. complications were found to be as striking as the difference in ratio of COX_1 and COX_2 inhibitory action as illustrated below.

- 5 The data as per the reference "Drugs of Today Vol. 32 Suppl. D1996 Page 5" as illustrated herein below shows that only four compound out of which Naproxen and Diclofenac are examples are selective for COX_2 with potency ratio < 1 . Among the many available NSAIDs, indomethacin and aspirin are COX_1 selective with potency ratio upto 100.

10	Drug	Cyclooxygenase 2/Cyclooxygenase1 ratio
	Piroxicam	250
	Tolmetin	175
	Acetylsalicylic acid	166
	Sulindac	100
15	Indomethacin	60
	Tolfenamic acid	16.7
	Ibuprofen	15
	Paracetamol	7.4
	Sodium salicylate	2.8
20	Flurbiprofen	1.3
	Carprofen	1
	Meloxicam	0.8
	Diclofenac	0.7
	Naproxen	0.6
25	Nimesulide	0.1

On the other hand while Naproxen has a comparative $\text{COX}_2/\text{COX}_1$ ratio (0.6) which is much lesser in comparison to other NSAIDs such as aspirin (166) and indomethacin (60).

- After careful and planned experimentation and with a quest to improve upon the drug candidates with relatively high therapeutic index and relatively high safety margin, the

inventors of the present invention conceived, planned and executed a research project to evaluate the zinc complexation on safety and efficacy of such drug candidates by taking examples of Naproxen and diclofenac.

5 The compositions of the present invention are therapeutically useful efficacious and safe, simple to manufacture and the process of manufacture is not time consuming. Besides the entire process and the drug manufactured thereby is not expensive.

SUMMARY OF THE INVENTION

10 The present invention discloses a novel pharmaceutical COX₂ selective composition having potency ratio less than 1 of Naproxen and/or one or more salts and/or adducts of Naproxen and zinc in one or more salt forms.

DETAILED DESCRIPTION OF THE INVENTION

The inventors have synthesised Zinc-diclofenac complex. This complex was found to have a more potent anti-inflammatory activity than diclofenac but it showed more toxicity in experimental animals.

15 In accordance with the present invention the pharmaceutical composition comprising Naproxen and zinc can be represented by the general formula (Drug)₂ Zn.nH₂O. In other words two molecules of the drug are reacted with one molecule of zinc in the presence of water.

20 The salts and adducts of Naproxen and Diclofenac are selected from the group optically pure Naproxen, Naproxen-Lysine salt, Naproxen-Cyclodextrin Adduct, Diclofenac Potassium, Diclofenac Diethylammonium, Diclofenac Epolamine and Diclofenac Tromethamine.

The salts of zinc are selected from zinc carbonate, zinc acetate, zinc sulphate, zinc chloride and zinc glucomate.

In accordance with another aspect of the present invention, is disclosed the synthesis of Naproxen-Zinc complex. After careful and planned experimentation on studies of pharmacological properties of the metal-ligands prepared it has been surprisingly found that Naproxen-Zinc complex have Synergistically higher anti-inflammatory and analgesic activities when compared to the parent drugs, Zinc alone and physical mixture of the drug and Zinc. There is a trend that as anti-inflammatory and analgesic properties of a NSAID compound increase, the side effects particularly the gastric-irritation also increases. The phenomena is attributed to inhibition of endogenous PEG₂ in gastric mucosa. Moreover it has been further surprisingly found that Naproxen-Zinc complex has least gastrototoxic side effects measured by ulcer index although it has a very high anti-inflammatory and analgesic activity.

In another embodiment of the present invention the composition is disclosed where the Naproxen is converted into a therapeutically safe and efficacious form i.e. Naproxen-zinc just before administration through oral or other construable routes. Such compositions are very simple to prepare, cost effective and useful. Such compositions have two parts. In one part Naproxen is present in a suitable pharmaceutical carrier and the other part contains a titrated reactive form of zinc with suitable pharmaceutical carrier. In such a composition both the parts are separately reacted in a suitable amount of medium or are present in a admixture form which is reacted just prior to administration.

The Naproxen-zinc complex was prepared by reacting Zinc Carbonate and Naproxen Sodium in primarily aqueous solvent, which may contain one or more polar organic solvent, in a molar ratio such that two moles of Naproxen react with one mole of Zinc. The complex thus formed was separated and dried.

The dried compound was subjected to elemental analysis, Infra-Red Spectroscopy Nuclear Magnetic Resonance spectroscopy and Differential Scanning Calorimetry (DSC). The formation of Naproxen Zinc complex was proved by comparing IR scans and DSC graphs of the complex with individual compound separately. Examination of the IR-spectra of Naproxen and the synthesised Zinc-Naproxen complex revealed a definite shift in absorption for the carboxyl group and the disappearance of carboxyl O-H stretching and bending. The shift occurred in the direction of longer wavelength, indi

cating that the carboxyl group of Naproxen is strongly involved in complexation with Zinc. Comparison of the H-NMR spectra of the two compounds did not show much difference as in the case of Naproxen due to exchange broadening involving the water in the solvent. The formation of the Zinc complex was further confirmed by C¹³-NMR spectra which revealed a strong shift in the absorption of -COOH and slight shifts in the case of other carbon atoms. DSC curves for Naproxen, the physical mixture of Naproxen and Zinc sulphate, and Zinc-Naproxen complex, showed that the endothermic peak of Naproxen disappeared completely in the Zinc complex with the appearance of two new endothermic peaks. The first peak is due to loss of water of crystallisation and the second sharper peak is due to melting with decomposition of the Zinc complex.

Diclofenac, Naproxen, Diclofenac-Zinc and Naproxen Zinc alongwith normal saline (as control) and Indomethacin (as standard Anti-inflammatory drug) were subjected to the pharmacological screening tests.

The anti-inflammatory of the compounds was tested by Carageenan - induced rat paw edema and ulcerogenic effect was evaluated by finding ulcer index.

The findings of the study are summarised in table 1 & 2 and depicted in fig. 1 & 2.

The invention provides Naproxen-Zinc complex and a process for manufacture thereof not reported earlier and the initial pharmacological testing indicates a strong possibilities that this complex may be developed as a therapeutically useful compound with reduced dry dose and better tolerability w.r.t gastric irritation.

PREPARATION OF ZINC-NAPROXEN COMPLEX

Naproxen (69.1 g, 0.3 ml) was dissolved in a solution of sodium bicarbonate (25.2 g, 0.3 ml) in water (500 ml) and filtered (pH 8.14). To this solubility was addeed slowly and with constant stirring a solution of zinc acetate dihydrate (35 g, 0.15 ml) in water (180 ml). Immediate precipitation occurred and the precipitates were filtered, washed with a minimum quantity of absolute alcohol and then with cold water, and dried under vaccum to a constant weight to give zinc-naproxen complex (yield, 69.75g, 83.1%); m.p. 222 - 224°C.

CHARACTERISATION OF ZINC-NAPROXEN COMPLEX

Comparison of the IR spectra of naproxen and its zinc complex showed the disappearance of Carboxyl OH stretching (3166cm^{-1}) in the IR spectrum of the complex occurrence of strong shift in the absorption due to C = O group stretching (1727 to 1689cm^{-1}) and the shift in -CH-COOH stretch from 1175 to 160cm^{-1} also indicating the possibility of host guest interchanging at this site. The displacement occurred in the directions of longer wavelength indicating that the carboxyl group of naproxen is strongly involved in complexation with zinc. Donation of electron to metal produces lower excitation states and therefore shifts to longer wavelengths (William et al., 1976)

The inventors effected the experiments to test the anti-inflammatory activity and toxicity by evaluating the ulcer index employing Naproxen-Zinc complex and Diclofenac- Zinc complex synthesised by them. It has been surprisingly found by the inventors that the anti-inflammatory activity of the Naproxen-Zinc and Diclofenac-Zinc complex were much more than the parent drug. However, Diclofenac Zinc complex suffered from the characteristic of being toxic, the toxicity being proportional to the increase in its anti-inflammatory activity.

The invention will now be described with reference to the following examples which are by way of illustration :

Example 1 : Tablet dosage form

Naproxen Zinc

equivalent to Naproxen 125 mg	152.25 mg
Lactose	- 229.15 mg
Starch	- 30.00 mg
PVP	- 10.00 mg
Micro Crystalline Cellulose	- 15.00 mg
Purified Talc	- 0.90 mg
Magnesium Stearate	- 12.00 mg
	300.00 mg

Example 2 : Tablet dosage form

Naproxen Zinc

	equivalent to Naproxen 250 mg	304.50 mg
	Lactose -	110.50 mg
5	HPMC -	30.00 mg
	Micro Crystalline Cellulose -	45.00 mg
	Purified Talc -	10.00 mg
		500.00 mg

Example 3 : Tablet dosage form

Naproxen Zinc

10	equivalent to Naproxen 375 mg	456.75 mg
	Micro Crystalline Cellulose -	148.25 mg
	HPMC -	30.00 mg
	Magnesium Stearate -	15.00 mg
		650.00 mg

Example 4 : Tablet dosage form

15 Naproxen Zinc

	equivalent to Naproxen 500 mg	609.00 mg
	Lactose (Directly compressible)-	141.00 mg
		750.00 mg

Example 5 : Tablet dosage form

Diclofenac Zinc

20	equivalent to Diclofenac 50 mg	56.85 mg
	Lactose -	90.00 mg
	Micro Crystalline Cellulose -	30.00 mg
	Polyvinylpyrrolidone -	7.15 mg
	Sodium Starch Glycollate -	16.60 mg
		200.00 mg

Example 6 : Capsule dosage form

Naproxen Zinc

equivalent to Naproxen 250 mg 304.50 mg

Micro Crystalline Cellulose - 133.00 mg

5 Magnesium Stearate - 7.00 mg

Sodium Lauryl Sulphate - 1.5 mg

500.00 mg

Filled in size '0' hard gelatin capsules shells.

Example 7 : Suspension dosage form

10 Naproxen Zinc

equivalent to Naproxen 125 mg 152.25 mg

Cane Sugar - 1000.00 mg

Glycerine - 200.00 mg

Sodium Saccharin - 4.00 mg

15 Dr. Sodium EDTA - 8.00 mg

Xanthan GUM - 100.00 mg

Pineapple Flavour - 4.00 mg

Purified Water q.s. to 5.00 ml

Example 8 : Enteric Coating Tablet

20 Diclofenac Zinc - 28.42 mg

equivalent to Diclofenac 25 mg

Lactose - 100.00 mg

Micro Crystalline Cellulose - 71.58 mg

25 HPMC phthalate - 36.00 mg

Isopropyl Alcohol* -
236.00 mg

*Lost in process

Example 9 : Chewable Tablet Dosage form

30 Naproxen Zinc - 456.75 mg

equivalent to Naproxen 375 mg

	Sugar Granular	-	365.25 mg
	Mannitol	-	100.00 mg
	Starch	-	60.00 mg
	Mixed Fruit Flavour	-	10.00 mg
5	Aspartame	-	2.00 mg
	Magnesium Stearate	-	6.00 mg
			1000.00 mg

Example 10 : Effervscent Tablet

	Naproxen	-	250.00 mg
10	Zinc Carbonate	-	80.00 mg
	Sodium Chloride	-	10.00 mg
	PEG 400	-	10.00 mg
			350.00 mg

Example 11 : Effervscent Tablet

15	Naproxen	-	500.00 mg
	Citric Acid	-	60.00 mg
	Zinc Carbonate	-	250.00 mg
	Sodium Chloride	-	20.00 mg
	PEG 400	-	20.00 mg
			850.00 mg

20

Example 12 Capsules dosage form

Naproxen-Zinc

	Cyclodextrin Complex	-	375.00 mg
	Lysine	-	115.00 mg
30	Magnesium Stearate	-	5.00 mg
	Sodium Lauryl Sulphate	-	5.00 mg
			500.00 mg

Fili in size '0' hard gelatin capsule shells

Example 13. Kit

Granules A

Naproxen	-	250.00 mg.
Citric Acid	-	30.00 mg.
5 PEG 400	-	20.00 mg.
		300.0 mg.

Granules B.

Zinc Acetate	-	130.00 mg.
10 Sodium Bicarbonate-		92.00 mg.
Sodium Chloride	-	28.00 mg.

Dissolve B in a glass of water and add A. Administer.

Example 14. Topical formulation

15 Naproxen Zinc	-	1% w/v
Preservative	-	0.005 to 0.5% w/v
Buffer/vehicle	-	98.5 to 98.995% w/v

The examples of the formulation should not be construed to limit the scope of
20 the invention. In fact following these examples, any desired pharmaceutical
formulation containing Naproxen and zinc can be prepared. The composition of
the invention can be in any form commonly employed for administration i.e drink

solution, a concentrated drink solution to be diluted before use, solution encapsulated in soft gelatin capsules, solution adsorbed on suitable adsorbents leading to formulations such as tablets, capsules, and granules, the solution freeze dried for oral, topical solution, or injectable dosage forms or the like.

- 5 Another embodiment of this invention is a kit which comprises one or more pharmaceutically acceptable doses of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and one or more acceptable doses of Zinc in one or more salt forms or a mixture thereof.

TABLE -1
Effects of administration of Indomethacin, Diclofenac Sodium, Naproxen, Naproxen-Zinc and Diclofenac-Zinc on Carageenan-induced paw edema in rats.

Group	n	Dose (mg/Kg i.p.)	0'	INCREASE IN PAW VOLUME (%)					
				15'	30'	60'	120'	180'	240'
Saline	3	-	27.54±13.72	34.98±6.03	28.9±8.85	57.94±8.7	66.96±8.4	137.08±34.24	133.34±9.50
Indomethacin	3	10	11.1±1.59	31.26±11.43	51.8±12.32	52.89±2.13	50.72±1.66	65.42±15.26	94.25±13.2
Diclofenac Sodium	3	5	13.9±7.15	30.75±6.46	36.71±5.48	41.17±8.65	54.29±3.89	109.6±23.27	101.53±8.89
Naproxen	3	4	14.77±6.36	23.64±10.05	39.12±4.02	38.93±9.05	69.67±10.01	84.22±13.6	99.05±14.7
Naproxen-Zinc	3	20	17.49±5	35.78±12.04	41.66±10.7	36.52±4.99	50.22±16.49	48.6±9.88	48.26±2.91
Diclofenac-Zinc	3	12	20.14±10.45	23.46±8.33	15.39±7.4	19.92±8.3*	25.97±11.07*	34.31±16.44	34.71±14.13

* P < 0.05

TABLE -2

Values of ulcer index obtained in each group of rats after administration of Indomethacin, Diclofenac Sodium, Naproxen, Naproxen Zinc and Diclofenac-Zinc

Group	n	Dose (mg/kg.p.o)	Ulcer Index
Saline	3	-	4.5±0.86
Indomethacin	3	10	09±2.5
Diclofenac Sodium	3	5	4.66±0.88
Naproxen	3	4	4.66±1.01
Naproxen-Zinc	3	20	4.16±0.44
Diclofenac-Zinc	3	12	8.16±0.83*

* P < 0.05

THE CLAIMS defining the invention are as follows:

1. A COX₂ selective pharmaceutical composition having potency ratio less than 1 comprising a complex of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -



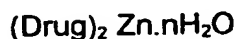
wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent.

2. A composition as claimed in claim 1 wherein the salts and adducts of
10 Naproxen are selected from the group comprising optically pure Naproxen, Naproxen-Lysine salt and Naproxen Cyclosporin adduct.

3. A composition as claimed in claim 1 wherein the salt forms of Zinc are selected from the group comprising Zinc carbonate, Zinc acetate, Zinc
15 sulphate, Zinc chloride and Zinc glucomate.

4. A composition as claimed in claim 1 which is in the form of a tablet, capsule, suspension, enteric coating tablet, chewable tablet, effervescent tablet and a topical formulation.

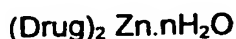
5. The use of an effective amount of a complex of Naproxen, its salt or adduct, or mixtures thereof, and zinc or a salt thereof, said complex represented by the formula -



- 5 wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent, in the preparation of a pharmaceutical composition for the treatment of an NSAID-indicated condition or symptom.

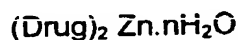
6. A method of treating an NSAID-indicated condition or symptom with reduced gastrointestinal side effects, said methods comprising:

- 10 administering an effective amount of a complex of Naproxen, its salt or adduct, or mixtures thereof, and zinc or a salt thereof, said complex represented by the formula -



- 15 wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent.

- 20 7. A process for the manufacture of a COX₂ selective pharmaceutical composition having potency ratio less than 1 comprising a complex of Naproxen and/or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -



wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent which comprises mixing Naproxen and/or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms



or mixtures thereof in the presence of water under conventional conditions of temperature and pressure.

8. A process as claimed in claim 7 wherein the salts and adducts of Naproxen are selected from the group comprising optically pure Naproxen, Naproxen-Lysine salt and Naproxen Cyclosporin adduct.
9. A process as claimed in claim 7 wherein the salt forms of Zinc are selected from the group comprising Zinc carbonate, Zinc acetate, Zinc sulphate, Zinc chloride and Zinc glucomate.
10. A process as claimed in claim 7 wherein Naproxen is dissolved in a solution of Sodium bicarbonate in water and filtered prior to mixing with a mixture of Zinc and salts thereof.
11. A process as claimed in claim 7 wherein Zinc in one or more salt forms or mixtures thereof are mixed with Naproxen and its salts or adducts or mixtures thereof under constant stirring.
12. A COX₂ selective pharmaceutical composition substantially as hereinbefore described with reference to any one of examples 1 to 14.

13. A method of treating an NSAID- indicated condition or system with reduced gastrointestinal side effects substantially as hereinbefore described with reference to Tables 1 and 2 and Figures 1 and 2.

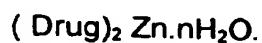
Dated this Nineteenth day of June 1998.

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ABSTRACT

A COX₂ selective pharmaceutical composition is disclosed. The composition has a potency ratio less than 1 and comprises a complex of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -



The drug in the above formula is Naproxen in a suitable pharmaceutical base/carrier or diluent.

FIG-1

Effects of administration of Indomethacin, Diclofenac Sodium, Naproxen, Naproxen-Zinc and Diclofenac-Zinc on Carageenan-induced paw edema in rats.

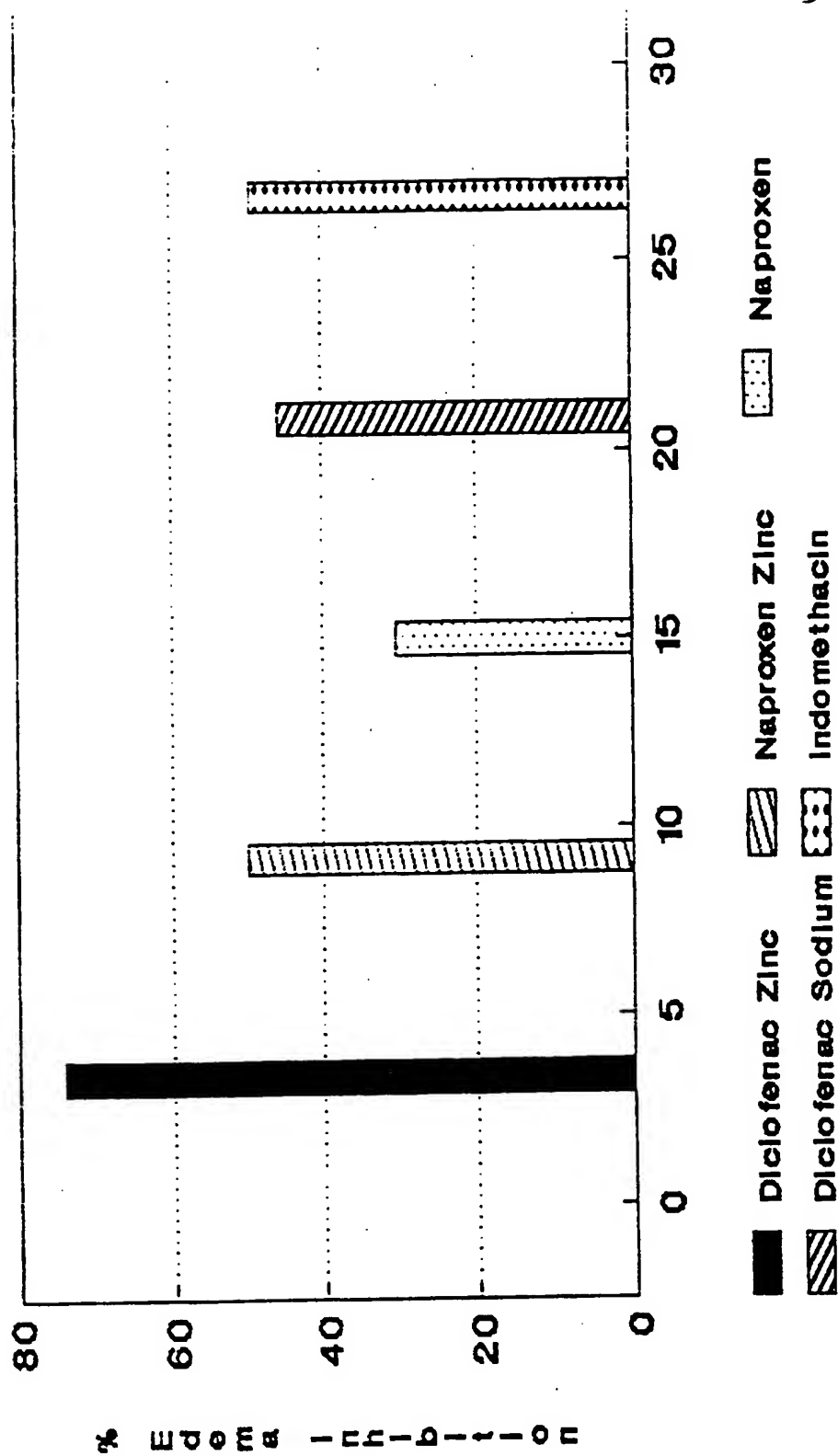


FIG-2
 Values of ulcer index obtained in each group of rats after administration of Indomethacin, Diclofenac Sodium, Naproxen, Naproxen Zinc and Diclofenac-Zinc

